

# Chapter 6 Chemical Bonds

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 that holds cations and  
 anions together. covalent.  
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 share a pair of electrons.  
 molecule. neutral group of  
 atoms that are joined  
 together by one or more  
 covalent bonds.  
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 Formulas 6.4 The  
 Structure of  
 Metals.Chapter 6-  
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 Covalent Bonding. A  
 covalent bond is a  
 chemical bond in which  
 two atoms share a pair of  
 valence electrons. When  
 two atoms share one pair  
 of electrons, the bond is  
 called a single  
 bond.Chapter 6 -  
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 is ionic, polar covalent, or  
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 on electronegativity  
 differenceChapter 6  
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 REVIEW Chemical Bonding  
 SECTION 1 SHORT  
 ANSWER Answer the  
 following questions in the  
 space provided. 1. a A  
 chemical bond between  
 atoms results from the  
 attraction between the  
 valence electrons and of  
 different atoms. (a) nuclei  
 (c) isotopes (b) inner  
 electrons (d) Lewis  
 structures 2. b A covalent  
 bond consists of (a) a  
 shared electron.6  
 Chemical BondingA  
 chemical bond in which  
 two atoms share a pair of  
 valence electrons. Define  
 polyatomic ion: A  
 covalently bonded group  
 of atoms that has a  
 positive or negative  
 charge and acts as a  
 unit.Name\_\_\_\_\_

\_\_\_\_\_ Chapter  
 6 Notes - Chemical  
 Bonding Chemical bond -  
 A mutual electrical  
 attraction between the  
 nuclei and valence  
 electrons of different  
 atoms that binds the  
 atoms together 6-1  
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 Bonding I. Types of  
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 Answers Pdfa chemical  
 bond resulting from the  
 sharing of an electron pair  
 between two atoms.  
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 covalent bond in which  
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 bonded atoms, resulting  
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 Covalent Bonding. The  
 attractions between the  
 shared electrons and the  
 protons in each nucleus  
 hold the atoms together  
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CHAPTER 6 REVIEW  
Chemical Bonding  
SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.  
Chapter 6 Chemical Bonds Section 6 2 Covalent Bonding ...  
a chemical bond resulting from the sharing of an electron pair between two atoms. nonpolar-covalent bond. a covalent bond in which the bonding electrons are shared equally by the bonded atoms, resulting in a balance of electrical charge.  
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Chapter 6: Chemical Bonds. STUDY. PLAY. ionic. type of bond that holds cations and anions together. covalent. bond in which two atoms share a pair of electrons. molecule. neutral group of atoms that are joined together by one or more covalent bonds. polar. [Chapter 6 Chemical Bonds Wordwise](#)

6.2 Covalent Bonding. The attractions between the shared electrons and the protons in each nucleus hold the atoms together in a covalent bond.

- Acovalent bond is a chemical bond in which two atoms share a pair of valence electrons.
- When two atoms share one pair of electrons, the bond is called a single bond.

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Bonding. A covalent bond

is a chemical bond in

which two atoms share a

pair of valence electrons.

When two atoms share one pair of electrons, the bond is called a single bond.

Chapter 2 Wordwise

Answer Key

Chapter 6 Notes -

Chemical Bonding

Chemical bond - A mutual electrical attraction

between the nuclei and

valence electrons of

different atoms that binds

the atoms together 6-1

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