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Stoichiometry b. Theoretically, how many moles of NH<sub>3</sub> will be produced? PROBLEMS Write the answer on the line to the left, Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N<sub>2</sub> are mixed with 12.0 mol of H<sub>2</sub> according to the ...

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