

Molecular Shapes Lab Activity Answers

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3: Lewis Structures and Molecular Shapes (Experiment ...

Molecule Shapes - VSEPR | Lone Pairs | Bonds - PhET ...

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Answers Name: _____ MOLECULAR SHAPES LAB ACTIVITY Purpose:

In this activity, you will be using a 3-dimensional simulator to

determine the patterns in molecular geometry and electron

geometry. When you are finished this lab, you will be able to

determine the shape of a basic molecule based on the bonds and

the lone pairs surrounding the central atom. [Molecular shapes](#)

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The molecular geometry main shapes are tetrahedral, trigonal planar, trigonal pyramidal, bent, and linear and are named by measuring the bond angles between the central atom and another atom bonded to it.

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Molecular Shapes The shape of a molecule depends on the distribution of atoms in space about the central atom, and their bond angles. Bond pair electrons and lone pair electrons repel one another, thus they will be arranged around a central atom as far apart as possible in order to minimize repulsions.

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Investigating Molecular Shapes with VSEPR The shape of a molecule will dictate many physical and chemical properties of a substance. In biological systems many reactions are controlled by how substrate and enzyme molecules fit together. Physical properties of substances, such as solubility and boiling point are also influenced by molecular geometry.

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